



A new type of 1,2,4-trioxanes structurally related antimalarial artemisinin

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Abstract

Co(II)-catalyzed triethylsilylperoxygenations of allylic alcohols 5 followed by acid catalyzed desilylative cyclizations with a pendant keto group furnished new types of 1,2,4-trioxanes 1-3 under very mild conditions in moderate yields. © 1999 Elsevier Science Ltd. All rights reserved.

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The discovery of artemisinin, an active principle of the Chinese medicinal plant qinghao, was an important milestone in antimalarial chemotherapy. Several classes of peroxides related to artemisinin and even synthetic 1,2,4-trioxanes were found to have potent antimalarial activities. Many synthetic methods for 1,2,4-trioxanes have been known, such as copper(II) catalyzed cyclization of vicinal hydroxy hydroperoxide, acid-catalyzed cyclization of hydroperoxy acetals with olefins or epoxides, 1,2-dioxetane with ketones or aldehydes, peroxyaldehyde with ketones or aldehydes, and cationic ring expansion of ozonides. Recently, we reported several bicyclic 1,2,4-trioxanes, which were synthesized from Co(II)-catalyzed oxygenation of cinnamyl alcohol followed by acid-catalyzed cyclization with various aldehydes or ketones. In continuing work on the synthesis of 1,2,4-trioxanes in the search for new antimalarial drugs, we wish to report new types of 1,2,4-trioxanes 1–3.

Pyrrolidine enamine of cyclohexanone was alkylated with acrylonitrile and then hydrolyzed to give 3-(2-oxo-cyclohexyl)propanenitrile (4). Addition of vinylmagnesium bromide to the ketone 4 gave 3-(2-hydroxy-2-vinylcyclohexyl)propanenitrile in a 5:1 *trans:cis* ratio. Then the nitrile was converted to the ketones 5a, 5b and 5c in 56, 39 and 62% yields, respectively (Scheme 1).¹⁰

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Scheme 1. Reagents: (a) CH₂=CHMgBr, THF, 0°C; (b) RLi, ether, rt; (c) Co(acac)₂ (0.1 equiv.), Et₃SiH (1.5 equiv.), ethanol, under O₂; (d) p-TsOH (cat.), CH₂Cl₂

Then Co(II)-catalyzed oxygenations with the double bonds of the compounds **5a**–**c** in the presence of triethylsilane yielded a 1:1 diastereomeric mixture of the triethylsilylperoxy alcohols **6a**–**c**, respectively. The triethylsilylperoxy alcohols **6a** and **6b** were cyclized by adding a catalytic amount of *p*-toluenesulfonic acid to give a 1:1 diastereomeric mixture of the corresponding 1,2,4-trioxanes **1** and **2** in 28 and 24% yields from the keto-allylic alcohols **5a** and **5b**, respectively. Since the two diastereomers of the **6c** have relatively different polarities, we were able to separate them using silica gel chromatography to give the isomers **7a** and **7b** in almost 1:1 ratio. Both diastereomers **7a** and **7b** were independently cyclized to give the 1,2,4-trioxanes **3a** and **3b** in 20 and 16% yields, respectively, as shown in Scheme 2. Note that all these 1,2,4-trioxanes **1**–**3** were structurally unique, in which two alkyl groups were substituted at both 3- and 6-positions of the 1,2,4-trioxane ring.

$$H_3C$$
 H_3C
 H_3C

We have studied 2D NMR experiments to accomplish ¹H and ¹³C NMR signal assignments and to confirm relative stereochemistry of the trioxanes **3a** and **3b** (Table 1). Proton J-networks and proton-carbon connectivities exhibiting overall covalent linkages of **3a** and **3b** were confirmed by ¹H, ¹³C NMR, DEPT, and HETCOR COSY and TOCSY experiments. The relative stereochemistry was finally confirmed by temperature dependent (-60 to 30°C) NOESY experiments with 150, 200, 250 ms mixing times. Direct NOE enhancements for **3b** were found between H¹³ and H^{5a} and between H¹³ and H⁵, while no such NOE for **3a** was found.

In summary, new 1,2,4-trioxanes 1–3, structurally related to antimalarial artemisinin, were synthesized by employing Co(II)-catalyzed triethylsilylperoxygenation and then acid catalyzed desilylative cyclization with a pendant keto group under very mild conditions in moderate yields. In view of the increasing demand for effective antimalarial 1,2,4-trioxanes, the present strategy would be of high value in searching for new antimalarials structurally related to artemisinin.

 $\label{thm:continuous} Table \ 1$ Proton and carbon chemical shift assignments for $\bf 3a$ and $\bf 3b$

	12 3 4 5 H 6 7 0 1 1 5a 8 2 0 10 9 8 3a	H ₃ C H H	12 3 4 5 H 6 7 0 1 1 5 a 8 2 0 1 1 9 3 8 3 b	E O H CH ₃ H
	¹ H (ppm)	¹³ C (ppm)	¹ H (ppm)	¹³ C (ppm)
3		101.5		102.0
4	1.92 (m)	30.45	1.93 (m)	30.02
5	1.35, 2.92	27.98	1.28, 2.68	26.43
5a	1.51 (m)	37.78	1.47 (m)	31.56
6	0.97, 1.67	34.11	1.22, 1.48	33.52
7	1.58, 1.75	29.51	1.35, 1.76	28.25
8	1.54	20.76	1.54	20.16
9	1.74	25.89	1.72	25.65
9a		70.84		72.05
10	3.70 (q, J = 6.4 Hz)	82.01	4.19 (q, J = 6.8 Hz)	83.75
12	1.25 (s)	25.49	1.26 (s)	25.62
13	1.44 (d, J = 6.4 Hz)	14.53	1.11 (d, $J = 6.8 \text{ Hz}$)	13.13

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- 11. Spectral data for **1** (1:1 diastereomeric mixture): ¹H NMR (400 MHz, CDCl₃) δ 4.04 (q, *J*=6.4 Hz, 1H for one isomer) and 3.57 (q, *J*=6.8 Hz, 1H for the other), 2.20–2.04 (m, 1H), 1.75–1.51 (m, 10H), 1.50–1.09 (m, 8H), 1.26 (d, *J*=6.8 Hz, 3H for one isomer) and 1.02 (d, *J*=6.4 Hz, 3H for the other), 0.90 (t, *J*=7.2 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 108.7, 108.6, 82.17, 81.88, 81.61, 78.65, 39.13, 38.22, 38.19, 33.43, 32.38, 30.72, 30.14, 29.68, 28.80, 28.14, 25.95, 25.72, 25.45, 25.33, 24.60, 24.31, 23.15, 23.11, 22.68, 22.49, 18.14, 14.22, 14.20, 13.54; FT-IR (neat, cm⁻¹) 2931, 2865, 1458, 1377, 1251, 1094, 972; HRMS (CI, CH₄) calcd for C₁₆H₃₁O₃(M+CH₅+): 271.2273; found: 271.2272. Spectral data for **2** (1:1 diastereomeric mixture): ¹H NMR (400 MHz, CDCl₃) δ 7.56 (m, 2H), 7.36–7.25 (m, 3H), 4.23 (q, *J*=6.4 Hz,

1H for one isomer) and 3.76 (q, J=6.8 Hz, 1H for the other), 2.38-2.22 (m, 1H), 2.06-1.91 (m, 1H), 1.90-1.70 (m, 4H), 1.70-1.50 (m, 3H), 1.50-1.38 (m, 1H), 1.35-1.16 (m, 3H), 1.35 (d, J=6.8 Hz, 3H for one isomer) and 1.02 (d, J=6.4 Hz, 3H for the other); ¹³C NMR (100 MHz, CDCl₃) δ 142.2, 142.0, 129.7, 128.1, 128.0, 126.7, 125.3, 125.2, 107.7, 107.5, 85.80, 82.75, 81.90, 79.33, 38.70, 33.02, 32.28, 31.38, 31.32, 30.64, 28.72, 28.06, 25.81, 25.59, 24.83, 24.51, 22.51, 22.35, 18.11, 13.33; FT-IR (neat, cm⁻¹) 2929, 2860, 1496, 1450, 1284, 1092; HRMS (CI, CH₄) calcd for $C_{18}H_{27}O_3(M+CH_5^+)$: 291.1960; found: 291.1962. Experimental procedure for the 1,2,4-trioxane 3a and 3b: To a solution of 5c (98.1 mg, 0.50 mmol) and Co(acac)₂ (17.8 mg, 0.050 mmol) in ethanol (10 mL) were added triethylsilane (120 µL, 0.75 mmol) at 0°C under oxygen atmosphere. The resulting solution was purged twice with an oxygen stream for 10 min and stirred under a slightly positive pressure of oxygen. After being stirred at room temperature for 10 h, the reaction solution was concentrated and then separated by silica gel chromatography with a 20:80 mixture of ethyl acetate and hexane to afford the two isomers (7a, nonpolar, 42 mg; 7b, polar, 35 mg, combined 45% yield). The isomers 7a and 7b were treated with p-toluenesulfonic acid (1.0 mg) in dichloromethane (10 mL) and stirred at room temperature for 2 h. Each reaction mixture was concentrated and chromatographed to afford the corresponding 1,2,4-trioxanes 3a (19.6 mg, 20%) and 3b (15.7 mg, 16%), respectively. Compound 3a: ¹H NMR (300 MHz, CDCl₃) and ¹³C NMR (75 MHz, CDCl₃) data are summarized in Table 1; FT-IR (neat, cm⁻¹) 2932, 2867, 1451, 1375, 1103. Compound 3b: ¹H NMR (300 MHz, CDCl₃) and ¹³C NMR (75 MHz, CDCl₃) data are summarized in Table 1; FT-IR (neat, cm⁻¹) 2931, 2863, 1457, 1376, 1246, 1109; HRMS (CI, CH₄) calcd for $C_{12}H_{21}O_3(M+H^+)$: 213.1491; found: 213.1497.